

1. IDENTIFICATION

Product Name Oxalic acid, dihydrate

Other Names No Data Available

Uses There are no uses advised against.

Chemical FamilyNo Data AvailableChemical FormulaC2H2O4.2H2O

Chemical Name Ethanedioic acid, dihydrate

Product Description No Data Available

Contact Details of the Supplier of this Safety Data Sheet

 Organisation
 Location
 Telephone

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40400 Shah Alam Sengalor, Malaysia

Emergency Contact Details

Chemcall

For emergencies only; DO NOT contact these companies for general product advice.

Organisation Location Telephone

New Zealand 0800-243622 +64-4-9179888

National Poisons Centre New Zealand 0800-764766

2. HAZARD IDENTIFICATION

Poisons Schedule (Aust) Schedule 6

Globally Harmonised System

Hazard Classification Hazardous according to the criteria of the Globally Harmonised System of Classification and Labelling of

Chemicals (GHS)

Hazard Categories Acute Toxicity (Oral) - Category 4

Acute Toxicity (Dermal) - Category 4

Redox Ltd Auckland Office NZBN 9429038410239











Serious Eye Damage/Irritation - Category 1

Pictograms





Signal Word Danger

Hazard Statements H302 Harmful if swallowed.

H312 Harmful in contact with skin.H318 Causes serious eye damage.

Precautionary Statements Prevention **P270** Do not eat, drink or smoke when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Response P312 Call a POISON CENTER or doctor if you feel unwell.

P330 Rinse mouth.

P302 + P352 IF ON SKIN: Wash with plenty of water.

P305 + P351 + P338 + IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses,

P310 if present and easy to do. Continue rinsing. Immediately call a POISON

CENTRE/doctor.

P362 + P364 Take off contaminated clothing and wash it before reuse.

Disposal P501 Dispose of contents/container in accordance with local / regional / national /

international regulations.

3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Oxalic acid, dihydrate	C2H2O4.2H2O	6153-56-6	<=100 %

4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed IF SWALLOWED: Rinse mouth, then drink plenty of water. Do NOT induce vomiting. Call a Poison Centre or

doctor/physician for advice. Never give anything by mouth to an unconscious person.

Eye IF IN EYES: Immediately flush eyes continuously with running water for several minutes, holding eyelids open and

occasionally lifting the upper and lower lids. Immediately call a Poison Centre or doctor/physician for advice. Remove contact lenses if present and easy to do. Continue flushing until advised to stop by a Poison Information Centre or a

doctor, or for at least 15 minutes.

Skin IF ON SKIN (or hair): Remove and isolate contaminated clothing and shoes. Immediately flush skin and hair with running

water for at least 15 minutes. Call a Poison Centre or doctor/physician for advice. Wash contaminated clothing and shoes

before reuse.

Inhaled IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Get medical

advice/attention. Give artificial respiration if victim is not breathing. Administer oxygen if breathing is difficult.

Advice to Doctor Treat symptomatically.

*Ensure that attending medical personnel are aware of the identity and nature of the product(s) involved, and take

precautions to protect themselves.

No information available.



Medical Conditions Aggravated by Exposure

5. FIRE FIGHTING MEASURES

General Measures If safe to do so, move undamaged containers from fire area. Cool containers with water spray until well after fire is out.

Flammability Conditions May burn but does not ignite readily.

Extinguishing Media Use dry chemical, Carbon dioxide (CO2), foam or water spray for extinction. Use extinguishing measures that are

appropriate to local circumstances and the surrounding environment.

Fire and Explosion Hazard Avoid generating dust; Fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is

a potential dust explosion hazard.

Hazardous Products of

Combustion

Fire may produce irritating, toxic and/or corrosive gases, including Carbon monoxide, Carbon dioxide, Formic aid.

Special Fire Fighting Instructions Contain runoff from fire control or dilution water - Runoff may cause pollution.

Personal Protective Equipment Wear positive pressure self-contained breathing apparatus (SCBA). Structural firefighters' protective clothing will only

provide limited protection.

Flash Point No Data Available
Lower Explosion Limit No Data Available
Upper Explosion Limit No Data Available

Auto Ignition Temperature >=400 °C

Hazchem Code No Data Available

6. ACCIDENTAL RELEASE MEASURES

General Response Procedure Ensure adequate ventilation. ELIMINATE all ignition sources (if dust clouds can occur). Do not touch or walk through

spilled material. Avoid generating dust. Avoid breathing dust and contact with eyes, skin and clothing.

Clean Up Procedures Collect up dry and deposit in waste containers for later disposal according to regulations (see SECTION 13).

*Keep the material dry if possible.

Containment Stop leak if you can do it without risk. Prevent dust cloud. Prevent entry into waterways, sewers, basements or confined

areas.

Decontamination Wash away remainder with plenty of water.

Environmental Precautionary

Measures

Prevent entry into drains and waterways. Any large spillage into watercourses must be alerted to the regulatory authority

responsible for environmental protection or other regulatory body.

Evacuation Criteria Spill or leak area should be isolated immediately. Keep unauthorised/unprotected personnel away.

7. HANDLING AND STORAGE

Handling Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. Ensure

adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Minimise dust generation and accumulation. Avoid breathing dust and contact with eyes, skin and clothing. Do not ingest. Use personal protective equipment as required (see SECTION 8). Dry powders can build static electricity charges when subjected to the friction of transfer and mixing operations. Provide adequate precautions, such as electrical grounding and bonding, or inert

atmospheres.

Storage Store in a cool, dry and well-ventilated place, out of direct sunlight. Keep container tightly closed. Avoid exposure to air

and moisture (hygroscopic). Keep away from heat and sources of ignition - No smoking. Keep away from food/feedstuffs

and incompatible materials (see SECTION 10).



Keep in the original container.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General For Oxalic acid (CAS No. 144-62-7):

Safe Work Australia Exposure Standard: TWA = 1 mg/m3; STEL = 2 mg/m3.
 New Zealand Workplace Exposure Standard: TWA = 1 mg/m3; STEL = 2 mg/m3.

- NIOSH REL: TWA = 1 mg/m3; ST = 2 mg/m3.

- OSHA PEL: TWA = 1 mg/m3.

- Immediately dangerous to life or health (IDLH) concentration: 500 mg/m3.

Exposure Limits No Data Available

Biological Limits No information available.

Engineering Measures A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust

ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing

dispersion of it into the general work area.

Personal Protection Equipment - Respiratory protection: In case of inadequate ventilation, wear respiratory protection. Recommended: Dust

mask/particulate respirator (refer to AS/NZS 1715 & 1716).

- Eye/face protection: Wear appropriate eye protection to prevent eye contact. Recommended: Do not wear contact

lenses. Tight fitting goggles with side shields or wide vision full goggles.

- Hand protection: Wear protective gloves. Recommended: Nitrile, neoprene, natural rubber, polyvinyl.

- Skin/body protection: Wear appropriate personal protective clothing to prevent skin contact. Recommended: Standard

work clothes, long pants, long sleeves, coveralls, safety shoes.

Special Hazards Precaustions No information available.

Work Hygienic Practices Do not eat, drink or smoke when using this product. Wash thoroughly after handling. Take off contaminated clothing and

wash it before reuse. Routine housekeeping should be instituted to ensure that dusts do not accumulate on surfaces.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical State Solid

Appearance Crystals or powder

Odour Odourless

Colour Uncoloured or white

pH ~0.7 (50 g/l)

0.0312 Pa (@ 25 °C) Vapour Pressure **Relative Vapour Density** No Data Available **Boiling Point** Sublimes at >160 °C **Melting Point** Sublimes at >160 °C **Freezing Point** No Data Available Solubility 108 g/L in water 25°C **Specific Gravity** 0.813 [EU A.3 method] **Flash Point** No Data Available

Auto Ignition Temp >=400 °C

Evaporation RateNo Data AvailableBulk DensityNo Data AvailableCorrosion RateNo Data Available

Decomposition Temperature >160 °C

DensityNo Data AvailableSpecific HeatNo Data Available



Molecular WeightNo Data AvailableNet Propellant WeightNo Data AvailableOctanol Water CoefficientNo Data AvailableParticle SizeNo Data Available

Partition Coefficient -1.7 (23 °C) [OECD Guideline 107]

Saturated Vapour ConcentrationNo Data AvailableVapour TemperatureNo Data AvailableViscosityNo Data AvailableVolatile PercentNo Data AvailableVOC VolumeNo Data Available

Additional Characteristics No information available.

Potential for Dust Explosion Avoid generating dust; Fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is

a potential dust explosion hazard.

Fast or Intensely Burning

Characteristics

No information available.

Flame Propagation or Burning Rate of Solid Materials

No information available.

Non-Flammables That Could Contribute Unusual Hazards to a

No information available.

Properties That May Initiate or Contribute to Fire Intensity

May burn but does not ignite readily.

Reactions That Release Gases or

Vapours

 $Fire/decomposition\ may\ produce\ irritating,\ toxic\ and/or\ corrosive\ gases,\ including\ Carbon\ monoxide,\ Carbon\ dioxide,$

Formic aid.

Release of Invisible Flammable

Vapours and Gases

No information available.

10. STABILITY AND REACTIVITY

General InformationThe substance in solution is a medium-strong acid. Reacts violently with oxidants causing fire and explosion hazard.

Reacts with silver compounds, forming explosive silver oxalate. Attacks some forms of plastic.

Chemical Stability Stable under normal conditions of use and storage.

Conditions to Avoid Avoid generating dust. Avoid exposure to air and moisture. Keep away from heat and sources of ignition.

Materials to Avoid Incompatible/reactive with alkalis, alkaline solutions, ammonia, acid chlorides, halogenates, oxidising agents, metals.

Hazardous Decomposition

Products

Fire/decomposition may produce irritating, toxic and/or corrosive gases, including Carbon monoxide, Carbon dioxide,

Formic aid.

Hazardous Polymerisation Hazardous polymerisation will not occur.

11. TOXICOLOGICAL INFORMATION

General Information

- Acute toxicity: Harmful if swallowed and in contact with skin. Corrosive on ingestion; May cause effects on Calcium balance. Signs of toxicity include nausea and vomiting, headaches, abdominal pain, diarrhoea, bloody stool, numbness and tingling of fingers and toes, muscular irritability, tetany, convulsions, shock, cardiac irregularities and circulatory collapse [NICNAS].
- Skin corrosion/irritation: Not irritating to skin. No skin irritation (Rabbit) [OECD TG 404].
- Eye damage/irritation: Causes serious eye damage. Irreversible effects on the eye (Rabbit) [OECD TG 405].
- Respiratory/skin sensitisation: Oxalic acid is not a skin sensitiser [OECD Guideline 429].
- Germ cell mutagenicity: Not considered to be genotoxic [NICNAS].
- Carcinogenicity: No evidence of carcinogenicity [NICNAS].



- Reproductive toxicity: Does not show specific reproductive or developmental toxicity [NICNAS].
- STOT (single exposure): Corrosion and irritant effects of the mouth and digestive tract, skin, eyes and respiratory tract have been reported following exposure to either the solid or concentrated solutions of oxalic acid [NICNAS].
- STOT (repeated exposure): May cause harmful cumulative effects (reduced thyroid function, renal toxicity, kidney damage/stone formation) following repeated oral exposure.

- Aspiration toxicity: No information available.

Acute

Ingestion Acute toxicity (Oral):

- LD50, Rat: >375 mg/kg bw. [Supplier's SDS].

Other Acute toxicity (Dermal):

- LD50, Rabbit: >20,000 mg/kg bw. [Supplier's SDS].

Carcinogen Category None

12. ECOLOGICAL INFORMATION

Ecotoxicity Aquatic toxicity:

- LC50, Fish (Leuciscus idus): 160 mg/l (96 h) [Supplier's SDS].

- EC50, Crustacea (Daphnia magna): 162.2 mg/l (48 h) [Supplier's SDS].

Persistence/Degradability Readily biodegradable.

Mobility No information available.

Environmental Fate Prevent entry into drains and waterways.

Bioaccumulation Potential No information available.

Environmental Impact No Data Available

13. DISPOSAL CONSIDERATIONS

General Information Dispose of contents/container via a licensed disposal company and in accordance with local/regional/national

regulations. Must not be disposed together with household garbage.

Special Precautions for Land Fill Contaminated packaging: Dispose of as unused product.

14. TRANSPORT INFORMATION

Land Transport (New Zealand)

NZS5433

Proper Shipping Name
Class
No Data Available
Subsidiary Risk(s)
No Data Available
No Data Available

UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.



Sea Transport

IMDG Code

Proper Shipping Name Oxalic acid, dihydrate No Data Available Class Subsidiary Risk(s) No Data Available **UN Number** No Data Available No Data Available Hazchem No Data Available **Pack Group Special Provision** No Data Available **EMS** No Data Available

Marine Pollutant No

Comments NON-DANGEROUS GOODS: Not regulated for SEA transport.

Air Transport

IATA DGR

Proper Shipping Name
Oxalic acid, dihydrate
No Data Available
Subsidiary Risk(s)
No Data Available
UN Number
No Data Available
Hazchem
No Data Available
Pack Group
No Data Available
Special Provision
No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for AIR transport.

15. REGULATORY INFORMATION

General Information OXALIC ACID

Poisons Schedule (Aust) Schedule 6

Environmental Protection Authority (New Zealand)

Hazardous Substances and New Organisms Amendment Act 2015

Approval Code Additives Process Chemicals and Raw Materials Subsidiary Hazard Group Standard 2020 HSR002503

*HSR003571 (Revoked)

National/Regional Inventories

Australia (AIIC) Listed

Canada (DSL) Not Determined

Canada (NDSL) Not Determined

China (IECSC) Not Determined

Europe (EINECS) 205-634-3

Europe (REACh) Not Determined



Japan (ENCS/METI) Not Determined

Korea (KECI) Not Determined

Malaysia (EHS Register) Not Determined

New Zealand (NZIoC) Listed

Philippines (PICCS) Not Determined

Switzerland (Giftliste 1) Not Determined

Switzerland (Inventory of Notified

Substances)

Not Determined

Taiwan (NCSR) Not Determined

USA (TSCA) Not Determined

16. OTHER INFORMATION

Related Product Codes OXACID1000, OXACID1001, OXACID1002, OXACID1003, OXACID1004, OXACID1005, OXACID1006, OXACID1007,

OXACID1008, OXACID1009, OXACID1010, OXACID1011, OXACID1012, OXACID1013, OXACID1014, OXACID1015, OXACID1016, OXACID1017, OXACID1018, OXACID1019, OXACID1020, OXACID1021, OXACID1500, OXACID1501, OXACID1502, OXACID1503, OXACID1510, OXACID1515, OXACID1800, OXACID1801, OXACID1802, OXACID1802, OXACID1803, OXACID2000, OXACID2001, OXACID2500, OXACID3000, OXACID3001, OXACID3500, OXACID4500, OXACID5000, OXACID5001, OXACID5002, OXACID5003, OXACID5004, OXACID5005, OXACID5006, OXACID5007, OXACID5008, OXACID5009, OXACID5010, OXACID5011, OXACID5012, OXACID5013, OXACID5014, OXACID5015, OXACID5016, OXACID5017, OXACID5018, OXACID5019, OXACID5020, OXACID5021, OXACID5022, OXACID5023, OXACID5024, OXACID5025, OXACID5026, OXACID5027, OXACID5028, OXACID5029, OXACID5030, OXACID5031, OXACID5032, OXACID5033, OXACID5034, OXACID5035, OXACID5036, OXACID5037, OXACID5038, OXACID5039, OXACID5040, OXACID5500, OXACID6000, OXACID6001, OXACID6002, OXACID5000, OXACID8000, OXACID8200, OXACID8300, OXACID8350, OXACID8400, OXACID8400, OXACID8500, OXACID8501, OXACID8510, OXACID8502, OXACID8503, OXACID

OXACID8850, OXACID8900, OXACID8925, OXACID9000, OXACID9500, OXACID9900

Revision 6

AICS Australian Inventory of Chemical Substances

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square CentimetresCO2 Carbon Dioxide

COD Chemical Oxygen Demand **deg C (°C)** Degrees Celcius

EPA (New Zealand) Environmental Protection Authority of New Zealand

deg F (°F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism **IDLH** Immediately Dangerous to Life and Health **immiscible** Liquids are insoluable in each other.

inHg Inch of Mercury inH2O Inch of Water

K Kelvin **kg** Kilogram



kg/m³ Kilograms per Cubic Metre

Ib Pound

LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50% (one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours.

LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one half) of a group of test animals.

Itr or L Litre

m³ Cubic Metre

mbar Millibar

mg Milligram

mg/24H Milligrams per 24 Hours

mg/kg Milligrams per Kilogram

mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component present.

mm Millimetre

mmH20 Millimetres of Water

mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health

NOHSC National Occupational Heath and Safety Commission

OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion

ppm Parts per Million

ppm/2h Parts per Million per 2 Hours

ppm/6h Parts per Million per 6 Hours

psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure

STEL Short Term Exposure Limit

TLV Threshold Limit Value

tne Tonne

TWA Time Weighted Average

ug/24H Micrograms per 24 Hours

UN United Nations

wt Weight

